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Computational Design of Electrocatalysts for Proton Exchange Membrane Fuel Cells

Proton exchange membrane fuel cell (PEMFC) has attracted enormous attention as a viable, clean, and sustainable power generation technology alternative to widely employed fossil fuel-based technologies. Hydrogen oxidation reaction at the anode of a PEMFC is facile. In contrast, oxygen reduction reaction (ORR) at the cathode of PEMFC is sluggish due to a strong O-O bond in O₂ molecule and requires an efficient electrocatalyst for the effective reduction of O₂ to H₂O. Currently, platinum group metals (PGM) are found to be the best ORR electrocatalysts. The price of rare and expensive Pt-based electrocatalysts contributes significantly to the total cost of a fuel cell and hinders its mass application. Therefore, it is of great interest to investigate how to reduce and even eliminate PGM contents in the ORR electrocatalysts. In this seminar, I will present a computational study elaborating the ORR reaction mechanisms on Pt alloy catalysts and carbon-supported nitrogen-derived non-precious transition metal (TM-N/C, TM=Fe or Co) catalysts. Specifically, I will first discuss how the sub-surface transition metal could cause a shift in the d-electron band of the Pt alloy catalysts (such as, PtNi, PtCo and PtFe) and thus favorably modify the energetics of the ORR occurring on these Pt alloy catalysts. Moreover, I will examine the possible pathways of the ORR on the TM-N₄ clusters of the TM-N/C catalysts using the first-principles density functional theory calculation method. In particular, I will show that the ORR activity of the TM-N₄ clusters could be tuned by tailoring the energy levels of the non-bonding d-orbitals of the central TM atoms. Consequently, I demonstrate in this seminar that physics-based computational techniques are essential for accelerating, achieving, and amplifying research discoveries in the current forefronts of developing electrocatalysts for PEMFCs.

Wednesday Seminar Series